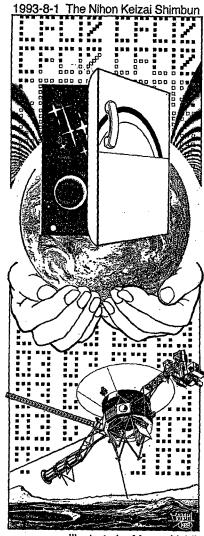


The International Thermoelectric Society

SCT-93

SHORT COURSE ON THERMOELECTRICS



Illustrate by Masami Ishli

FOR THE GREEN 21st CENTURY

Pacific Convention Plaza Yokohama Japan

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Short Course on Thermoelectrics - 1993 (SCT-93)

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SCT-93 Short Course on Thermoelectrics Nov. 8, 1993

An Overview of Thermoelectricity by Prof. David D. Allred Brigham Young University Provo, UT 84602 USA Philosophy of this talk: History and review

History because it is interesting, fairly easy to grasp and a good way to introduce the subjects.

Review because we come from a variety of disciplines and backgrounds. It is good to come together before we start the more rigorous parts of the course. This can provide the comfort that you will be able to understand what is coming.

An Outline:

- I. Introduction
 - A. Some Triumphs:
 - 1. Remote power supplies for outer space& terrestrial appli.
 - 2. Cooling. Both commercial and scientific/technological
 - B. History of TE:
 - 1. Discovery of Thermoelectric effects.
 - a. 1821- Seebeck- The symbol S, α
 - b. 1833- Peltier-temp changes- Π
 - c. 1838- Lenz explains.
 - 2. Thermodynamics is explored.
 - a. Lord Kelvin-The interrelation of S, Π
 - b. 1885- Rayleigh- power generation?
 - c. 1911- The discovery of Figure of Merit, Z. Altenkirch
 - d. A discussion of notation (can vary):

- 3. Modern period- Semiconductors.
 - a. "Minerals vs Metals"- The importance of high S.
 - b. 1930's-Synthetic semiconductors
 - c. 1947 Telkes- a 5% generator.
 - d. 1949 Ioffe- Theory of Semiconductor TE.
 - e. 1953 first refrigerator.
 - f. 1956 Ioffe's idea of alloying.
- 4. Applications & Optimism- Early '60's.
 - a. Space applications of generators.
 - b. Solid State was magic word.
- 5. Reassessments.
 - a. The stubborn practical limit of ZT=1 niche markets.
 - b. Oil Embargo.
 - c. Government-Star wars and SP-100.
 - d. Commercial applications of cooling.
- 6. Will there be ZT>>1? The future.



Fig. 4. SP-100 Class Reactor Powering an Electric Propulsion Spacecraft Past Io

C. History of TE:

- 1. Discovery of Thermoelectric effects.
 - a. 1821- Seebeck discovered that a compass needle would be deflected when placed near a closed loop made of two dissimilar metal when one of the junctions was heated. Was this then a magnetic phenomena? The formulation of the Seebeck series. A voltage difference can be found which depends on the temperature and the pair of materials used. $\Delta V = S \cdot \Delta T$. We will talk as though the properties of single materials, rather than pairs, were available.

S or α has units of $\mu V/K$ Metals up to about 10 $\mu V/K$ Semiconductors > 200 $\mu V/K$ b. 1833- **Peltier** announced that there are temperature changes at a junction of dissimilar metals when current is caused to flow.

$$Q = \prod i$$

Here Q is the rate at which heat is absorbed or rejected, i is the electrical current and Π is the Peltier coefficient.

c. 1838- Lenz explains that heat is either absorbed or released at a junction depending on the direction of current flow.

- 2. Thermodynamics is explored.
 - a. Lord Kelvin- relates Seebeck and Peltier coefficients and announces a **third** TE effect: The **Thompson effect**-Heating or cooling of a single, homogeneous conductor in a temperature gradient. $\Pi = S \bullet T$

- b. 1885- Rayleigh considers using Seebeck effect for the generation of electricity.
- c. 1911- The discovery of Z. Altenkirch creates a satisfactory theory of thermoelectricity for power generation and cooling. There needs to be high S, low thermal conductivity and high electrical conductivity.

A discussion of notation (which can vary):

There are three things:

- 1. Power factor = $S^2 \sigma$
- 2. Figure of Merit, $Z=S^2\sigma/\kappa$ has units of inverse temp: 1/kelvin
- 3. Dimensionless Figure of Merit, ZT

Heat conduction is factor in Figure of Merit, $Z=\alpha^2 \sigma/\kappa$ Thermal conductivity is represented as k, K, κ (kappa) or λ (lambda) and has units of watts/kelvins•meters. (There are other important quantities like thermal resistivity, W=1/ κ and thermal diffusivity.) I will mix symbols.

Typical values of κ are:

100 to 300 W/Km or 1 to 3 W/cmK for metals. Electrons are also carrying heat. Wiedemann Franz law. $\kappa/\sigma=L$ Lorentz constant. This was evidence for the free electron theory of metals.

~2 to 8 W/K•m for TE grade semiconductors but,

90 to 1000 W/Km for pure diamond lattice solid like Ge, Si, diamond. Stiff lattices conduct heat with vibrations like sound called longitudinal phonons k=kcarriers+klattice

 $Z=S2\sigma/\kappa$ = up to ~10-3 for very good TE at their maximum.

WARNING: S, σ , κ are interrelated. If usually $\sigma \uparrow$, S \downarrow . $\kappa \downarrow$, $\sigma \Downarrow$

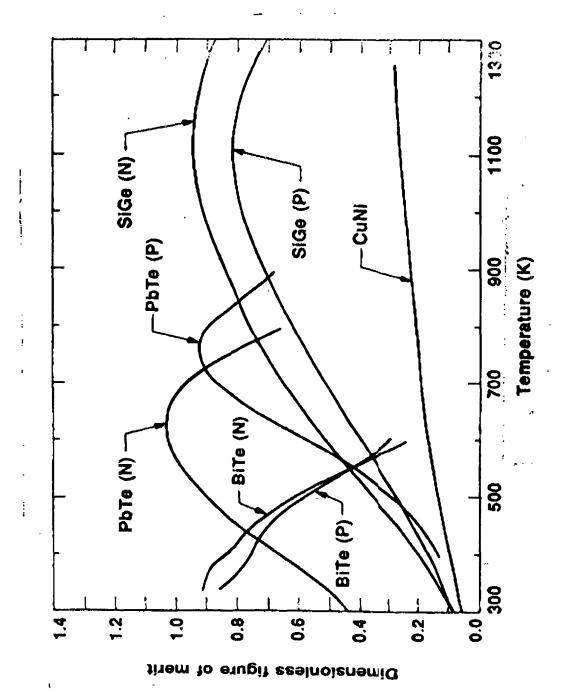
- 3. Modern period- Semiconductors.
 - a. "Minerals vs Metals"- The importance of high S.

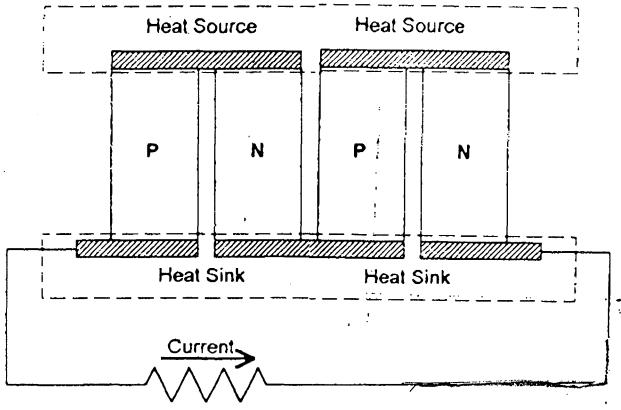
metals	advantages malleable, properties relatively constant, stable	•	uses thermocouple wire
semiconductor (heavy doped)	high S, moderate Z	brittle, temp. depend properties, unstable	TE devices, composed of small modules.

Many of the technological difficulties of TE comes out of the fact that they are semiconductors not flexible metals.

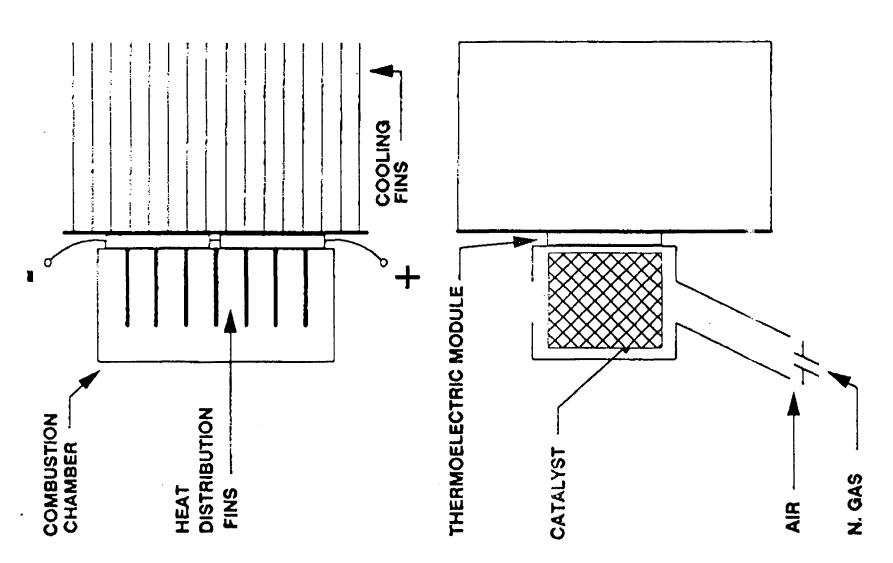
- b. 1930's-Synthetic semiconductors are studied.
- c. 1947 Telkes- a 5% generator.

Some facts- Balance of system also must be figured into the system calculations, weight and expense. **Presentations** # 5-8





Schematic of typical thermoelectric power generator. tion is achieved by replacing the resistor with a current rad reversing the current flow.



TELEDYNE THRMOELECTRIC POWER ASSEMBLY FIG. 1.1

Figure 15. Water Box Module Exploded View

d. 1949 Ioffe- Theory of Semiconductor TE.

See 2nd & 4th presentations. These are heavily doped: 0.1 to several % of atoms contribute carriers. Most are low to moderate band gap and involve elements of higher atomic number. Many are soft or brittle, covalent, somewhat unstable or reactive compounds of low or moderate melting points.

e. 1953 first Refrigerator or Heat Pump.

Diagram of cooler

Note: Thermoelectrics are not heat annihilators-- A story Discussion of COP

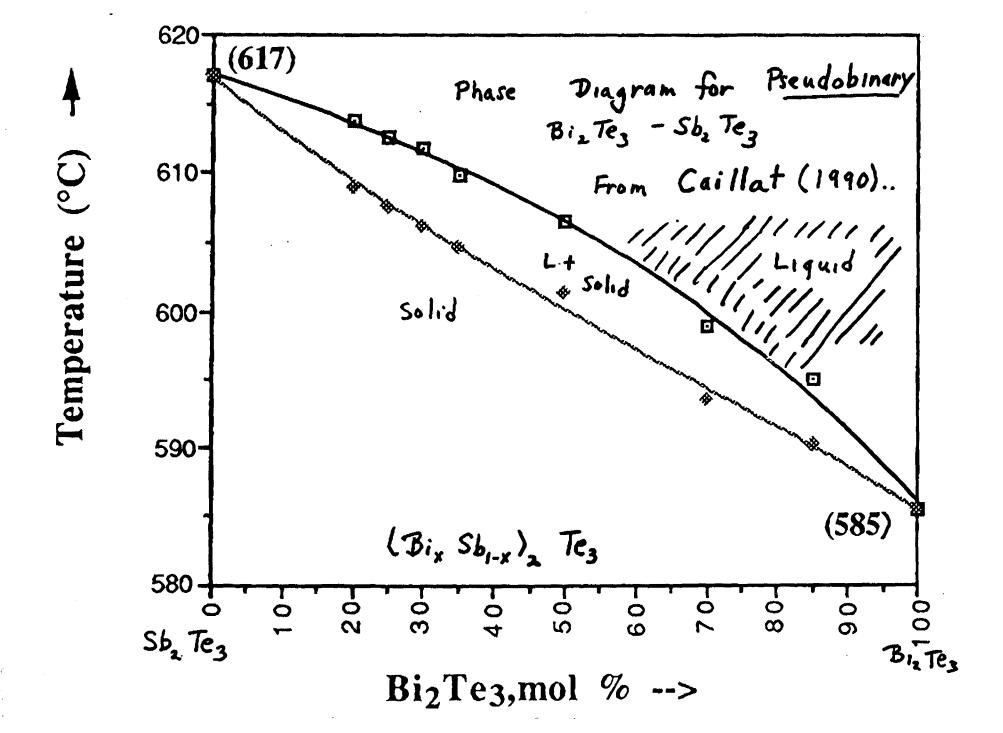
f. 1956 Lowering κ Ioffe's idea of alloying of isomorphic compounds to lower thermal conductivity. What is heat conduction? What are the mechanisms of heat conduction? (see presentation #2)

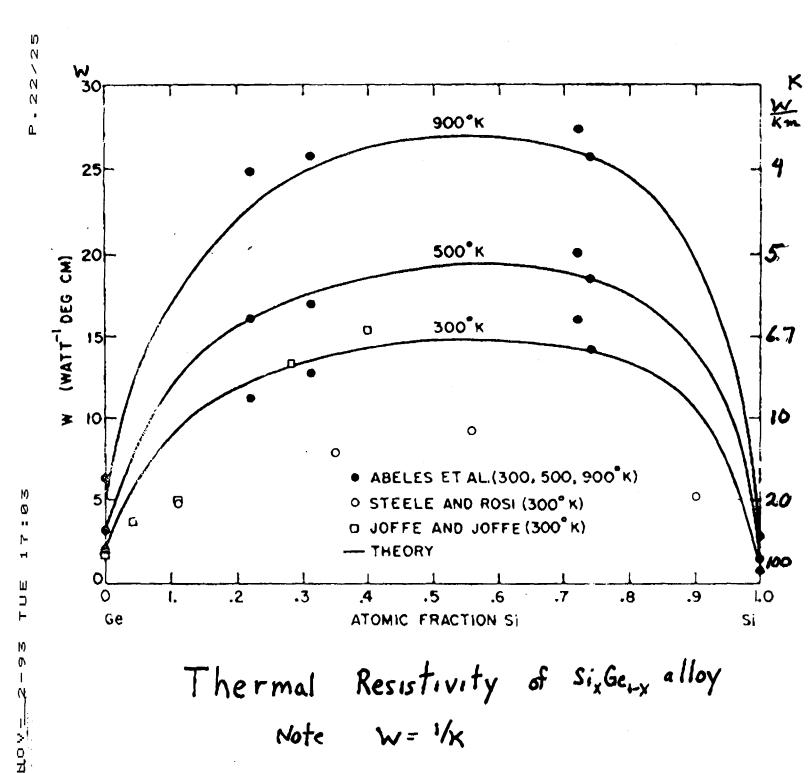
The role of current carriers: Wiedemann Franz law. κelectronic is proportional to σ.

*Microscopic view of solids as a lattice. What is a phonon? Phonon dispersion curve.

Imperfections in crystals

- a) carriers
- b) phonons
- c) variations in atoms-point defects-charge, mass
- d) extended defects...Grain boundaries





- 4. Applications and Optimism- Early 1960's.
 - a. Space applications of generators.
 - b. Solid State was magic word and there were projections of replacing many common, fluid-based thermodynamic cycles for terrestrial power and refrigeration.

- 5. Reassessments.
 - a. The stubborn practical limit of ZT=1 $\rightarrow \phi = \eta/6$, \Rightarrow niche markets. (see presentations #3, 4, 5 & 8)
- Max Power output: generators $\phi \approx \eta/[2.0 + 4/ZT]$, where $\eta = (T_h-T_c)/T_h$, which is the carnot efficiency. This is for cold side about equal to hot side temp; if cold it much lower than hot the constant in denominator tends to 1.5.

Max effic. : generators

$$\phi = \gamma \eta = \eta [\sqrt{1 + ZT_{av}} - 1]/[\sqrt{1 + ZT_{av}} + T_c/T_h]$$

Coefficient of Performance (COP) of coolers is best given graphically but often less than 1. $\epsilon = \frac{1}{\tau} - \frac{1}{\tau}$

fig 2.5 if possible

6. Will there be ZT>>1? The future.

A ZT of 2 would give generators with $\eta_{\text{max pow}} \sim 1/4$ of Carnot and $\eta_{\text{max eff}} = 0.3$ of Carnot. It would provide a cooler with 1/4 of Carnot.

A ZT of 3 would give generators with $\eta_{max\ pow} = 1/3$ of Carnot and $\eta_{max\ eff} = 0.3$ to 0.4 of Carnot. It would provide a cooler with 1/3 of Carnot.

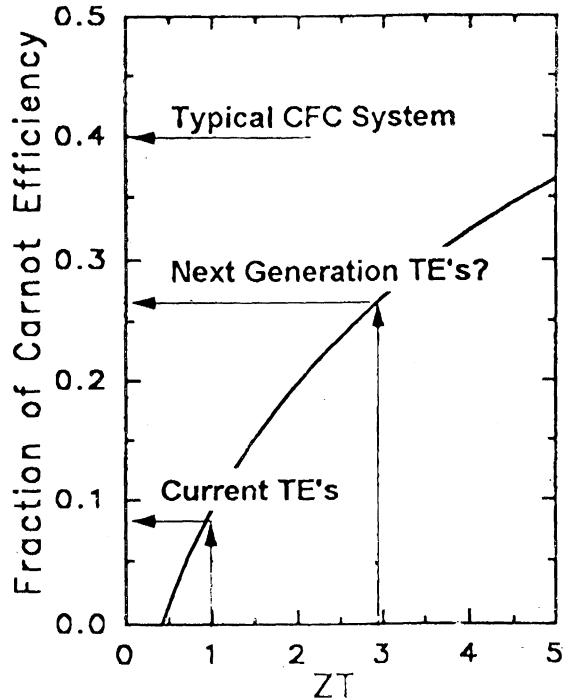


Figure 3. Effect of the Thermoelectric Figure of Merit on the Performance of a Thermoelectric Cooler.